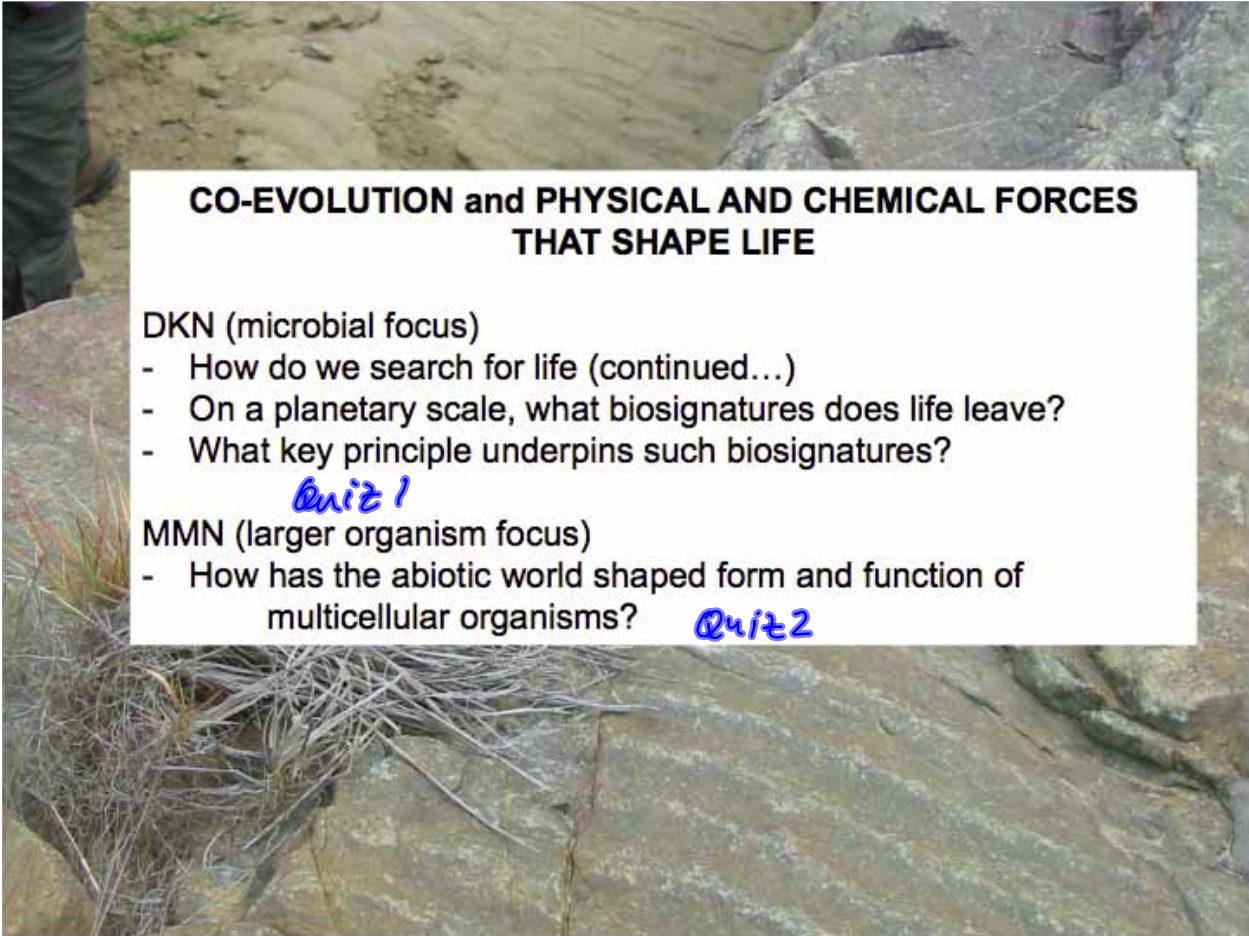




CO-EVOLUTION and PHYSICAL AND CHEMICAL FORCES THAT SHAPE LIFE

DKN (microbial focus)

- How do we search for life (continued...)
- On a planetary scale, what biosignatures does life leave?
- What key principle underpins such biosignatures?

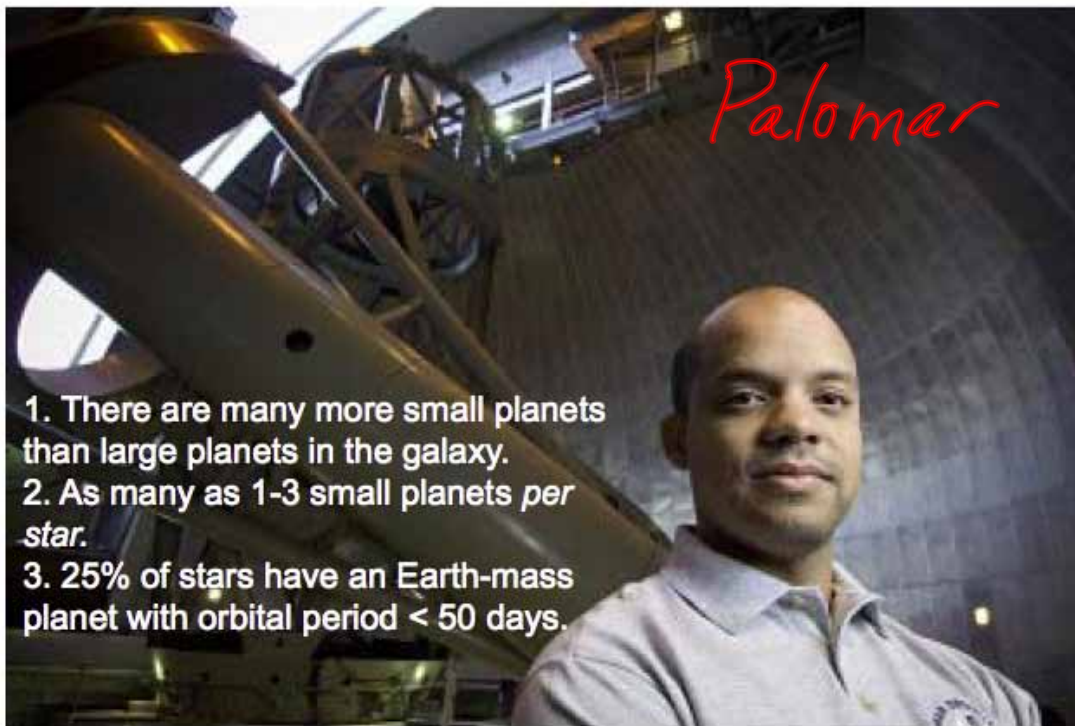
Quiz 1

MMN (larger organism focus)

- How has the abiotic world shaped form and function of multicellular organisms?

Quiz 2

Prof. John Johnson is on the hunt for Exoplanets



Can we define generic "biosignatures" that could help us search for life on other planets? What would you look for if you were the chief scientist?

nucleic acids, structured mineral deposits, H_2O

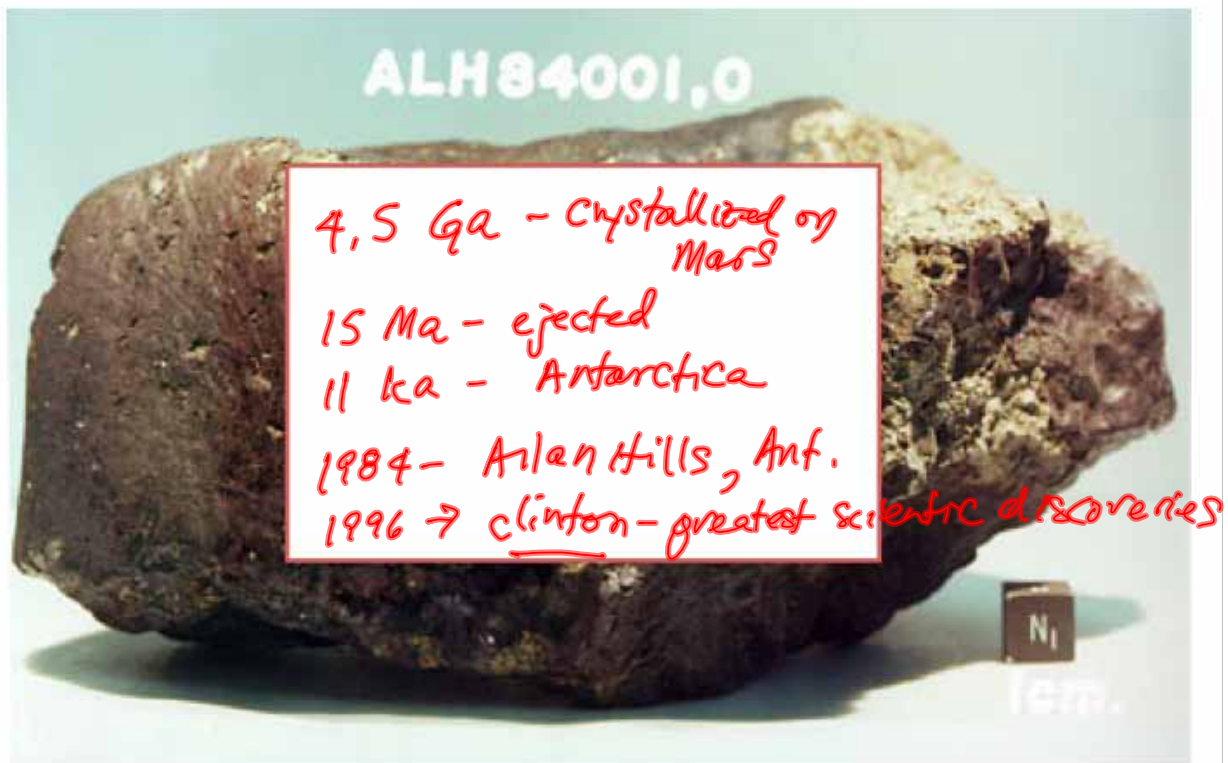
Mars Science Laboratory (MSL) "Curiosity" lands Aug. 5, 2012



Sometimes Mars comes to us, so we have a head start...



Sometimes Mars comes to us, so we have a head start...



Search for Past Life on Mars: Possible Relic Biogenic Activity in Martian Meteorite ALH84001

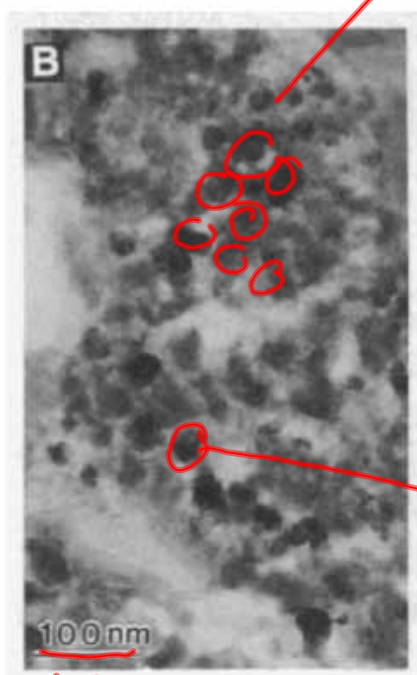
David S. McKay, Everett K. Gibson Jr.,
Kathie L. Thomas-Keprta, Hojatollah Vali,
Christopher S. Romanek, Simon J. Clemett,
Xavier D. F. Chillier, Claude R. Maechling, Richard N. Zare

Fresh fracture surfaces of the martian meteorite ALH84001 contain abundant polycyclic aromatic hydrocarbons (PAHs). These fresh fracture surfaces also display carbonate globules. Contamination studies suggest that the PAHs are indigenous to the meteorite. High-resolution scanning and transmission electron microscopy study of surface textures and internal structures of selected carbonate globules show that the globules contain fine-grained, secondary phases of single-domain magnetite and Fe-sulfides. The carbonate globules are similar in texture and size to some terrestrial bacterially induced carbonate precipitates. Although inorganic formation is possible, formation of the globules by biogenic processes could explain many of the observed features, including the PAHs. The PAHs, the carbonate globules, and their associated secondary mineral phases and textures could thus be fossil remains of a past martian biota.

Science, 1996

- Abiotic processes?

What do you think?



160 nm

clump

=
?



chain

50 nm single domain

Extra credit challenge (worth 2 correct iClicker questions)

What configuration of magnetite (chains or clumps) do you think is energetically preferred in the absence of life? (*i.e.* what would happen to the magnetite in *M. magneticum* if you dissolved its membranes and proteins?)

Based on what you have learned in Ph 1c, discuss qualitatively and/or quantitatively. ✓

How does this affect your thinking about whether the magnetite in ALH8001 is a robust biosignature? 

You can ask TAs for general guidance and collaborate with your fellow students, but what you turn in must be in your own hand.

Turn in at the beginning of class next Tuesday.



Life Detection: Let's ground-truth this discussion with the Earth

What features of Earth that are linked to life (i.e. necessary for it or a product of it) that you can see from space?



$H_2O \rightarrow$ pre-reg. ↙

4/3

- Green] chlorophyll

- O_2

biogenic trace

$\rightarrow N_2O$] abiotic
can
 CH_4] make

$\rightarrow$ concentration

liquid $\xrightarrow{\text{hyothermal}}$ / FeS

Why is water good for life?

- Dissolves most things

- Solvent/liquid?

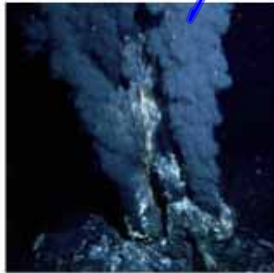
$\rightarrow$ app. rates for biochemistry

C-life: covalent bond formation
H-bonds

- Floats when freezes $\rightarrow$ expands

- High heat capacity $\rightarrow$ regulates T
compatible seasons, diurnal

- Large liq. range (pure H_2O $0^\circ\text{C} - 100^\circ\text{C}$
+ salt $\downarrow$ freezing T
 $\uparrow$ pressure $\uparrow$ boiling T

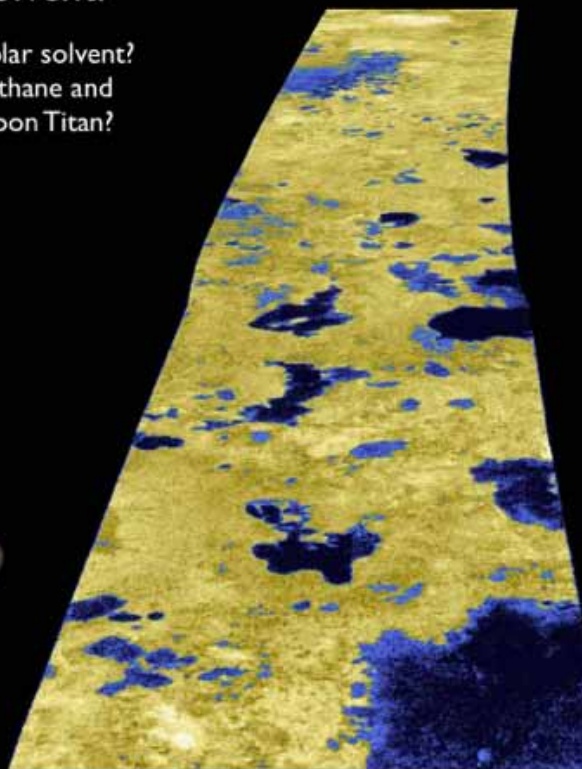


Prospects for 'Weird Life'?

Water = Polar solvent.

Could life 'work' in a non-polar solvent?
For instance, the liquid methane and ethane lakes of Saturn's moon Titan?

* ← Titan

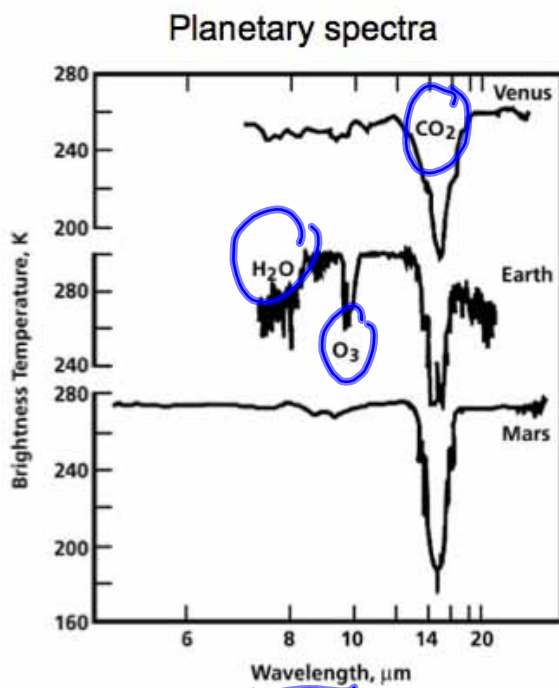


Huygens lander image of what may have been a methane river basin. The rocks of water ice are ~10 cm across.

False color Radar image of Titan's surface showing methane/ethane lakes.

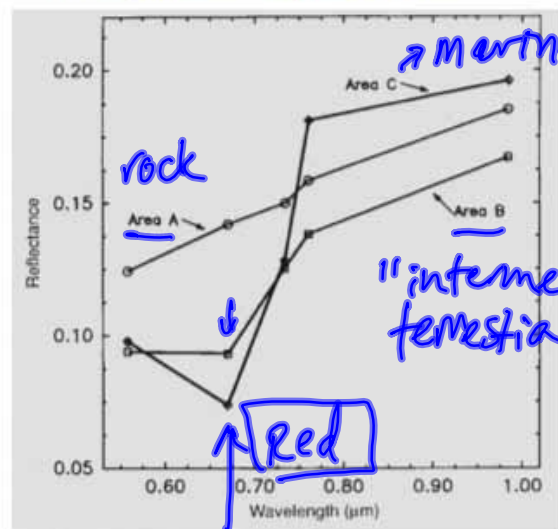
Images: NASA/JPL/ESA

Life Detection: Let's ground-truth this discussion with the Earth



Prof. Geoff Blake (CCE, GPS)

Chlorophyll absorption in red

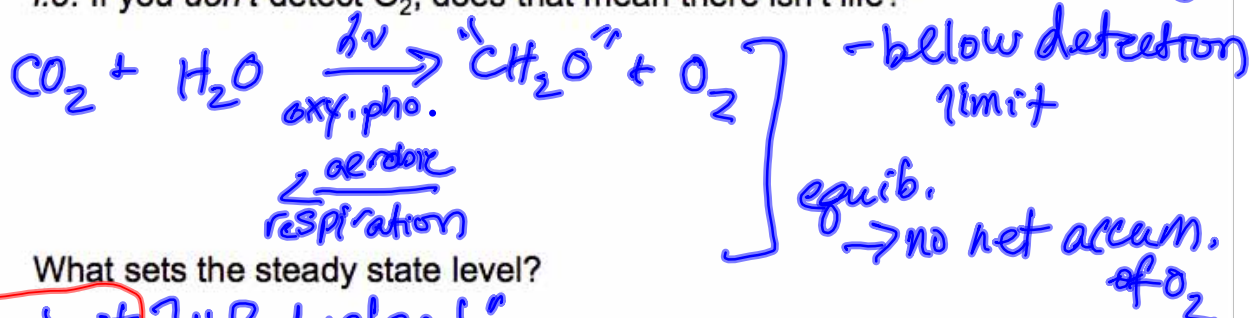


Sagan 1993

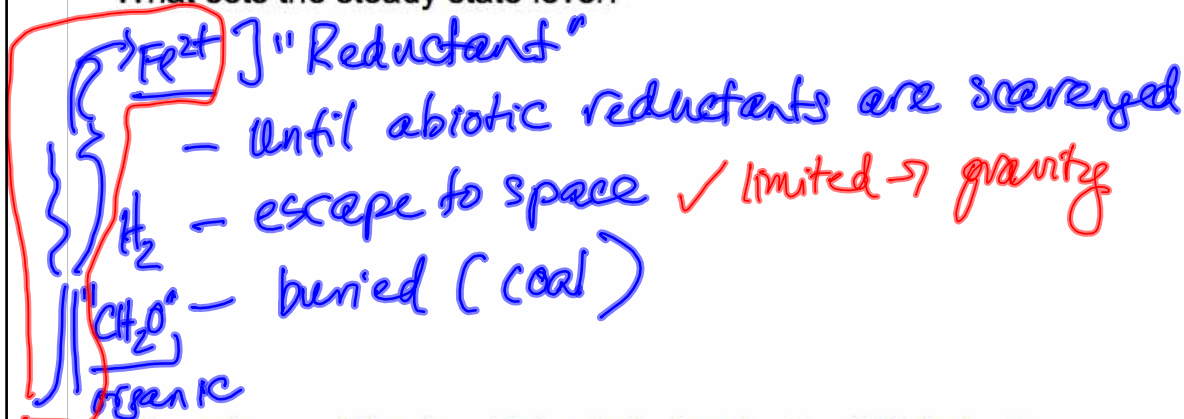
NASA SeaWiifs Movie

Is O_2 a Robust Biosignature?

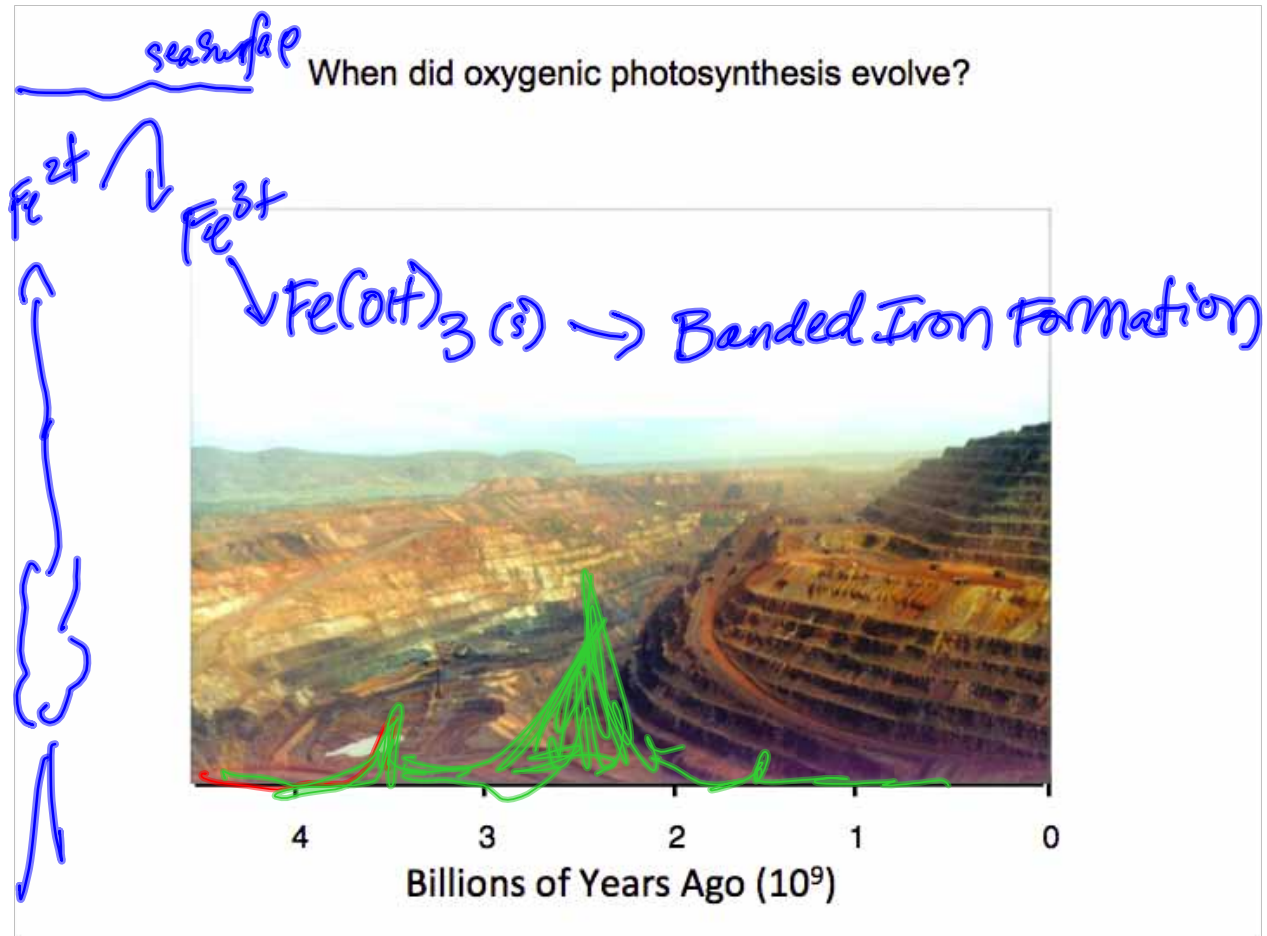
i.e. If you *don't* detect O_2 , does that mean there isn't life?

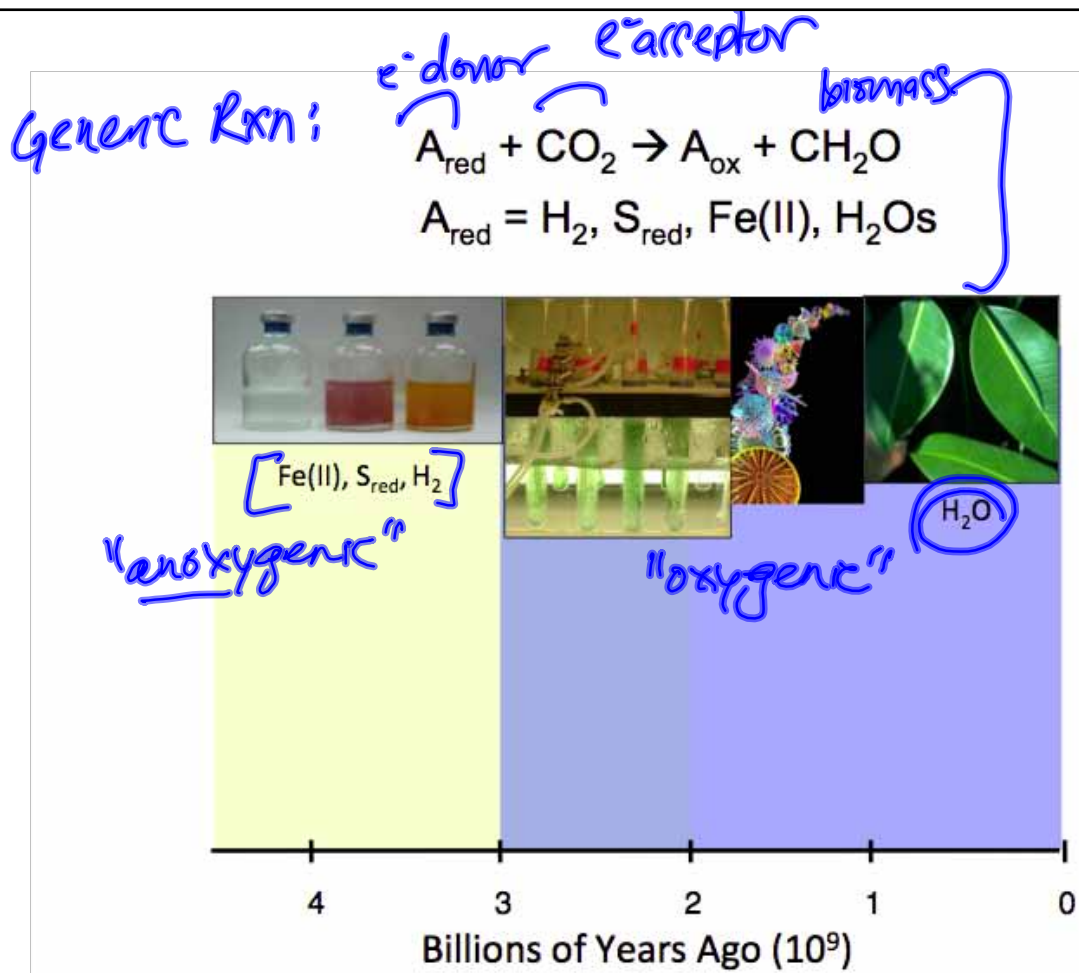


What sets the steady state level?

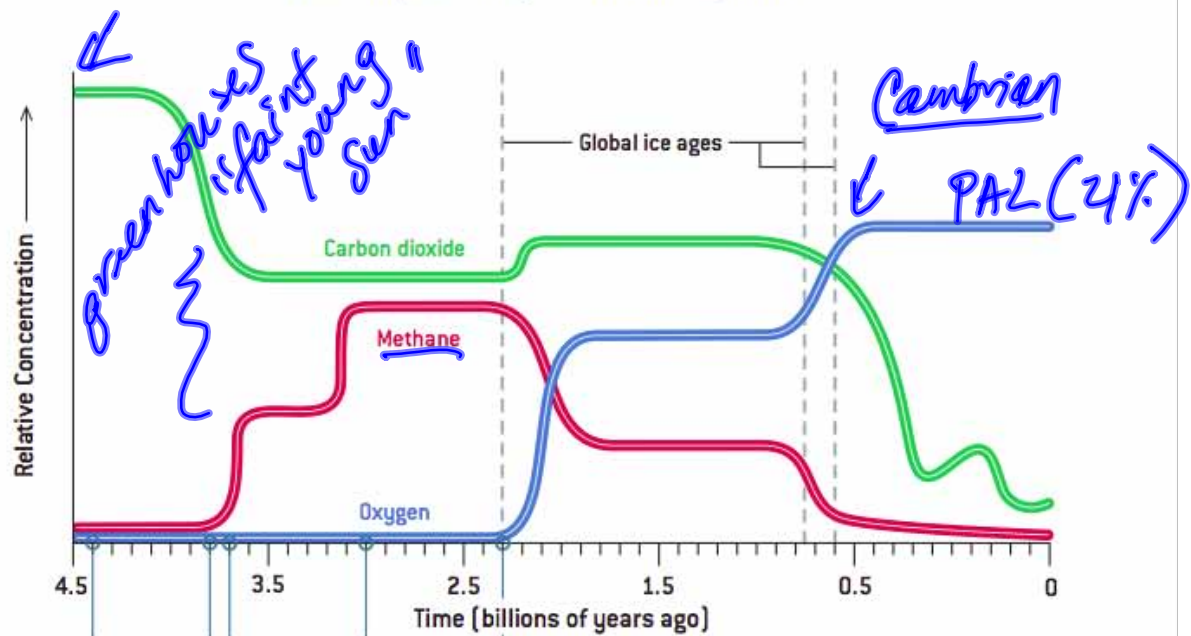


Complex combination of physical, chemical and biological processes set the planetary redox state. This can (and has!) changed through time.

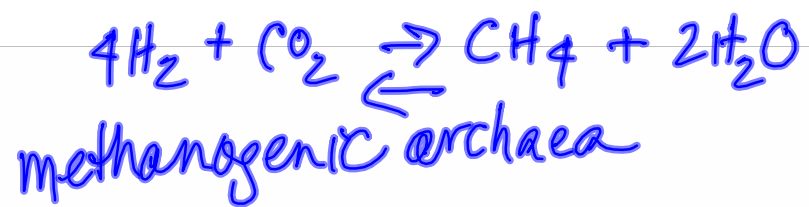




When did O_2 accumulate to the present atmospheric level (PAL)?
Was CH_4 an early Earth biosignature?

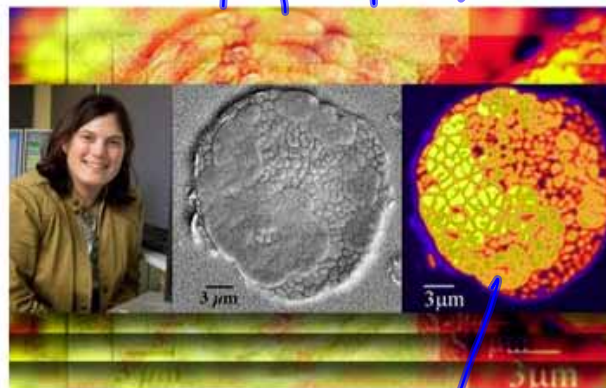
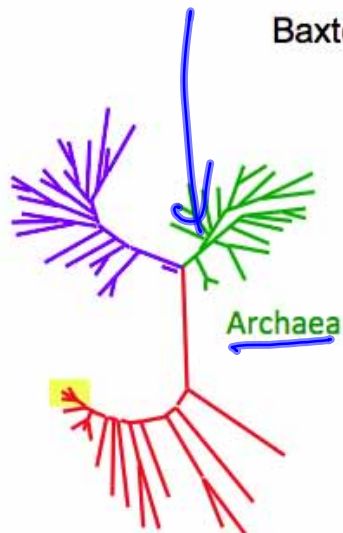


Kasting, Scientific American



Baxter Pond Volta Flame Movie

Prof. Orphan



week 8

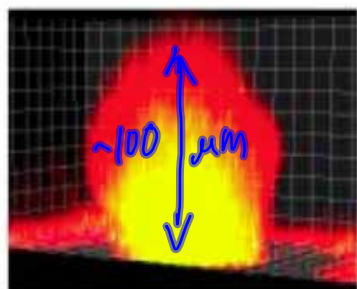
symbiotic
other bacteria

Concept of BIOGEOCHEMICAL DIS-EQUILIBRIUM: predictable layered sequence of metabolites

→ Kill/poison → layers disappear
 → ∴ biological consumption/production of metabolites outpaces diffusion
 μ-meters → centimeters

→ Kinetic fingerprint

diatoms Black Sea



Nealson and Conrad (



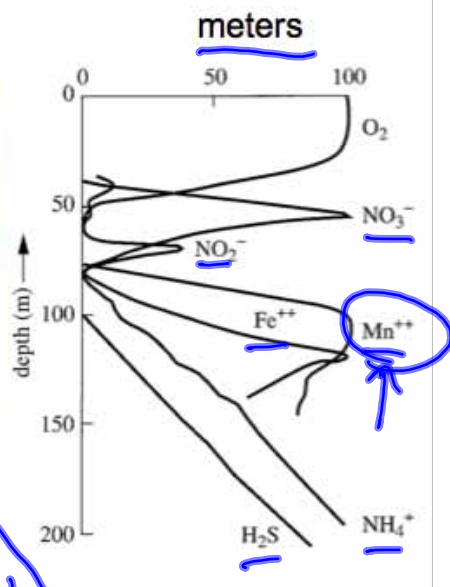
microbial mat

purple bacteria

→ anoxy. photo

cyano

Sulfate-red



BIOGEOCHEMICAL DISEQUILIBRIUM plays by thermodynamic rules

→ Redox couples:

A_{red}/A_{ox}

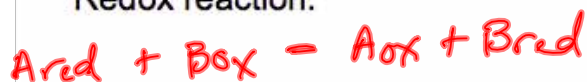
vs.

A_{ox}/A_{red}

oxidation
(lose e^-)

reduction
(gain e^-)

Redox reaction:



e^- donor e^- acceptor
reductant oxidant

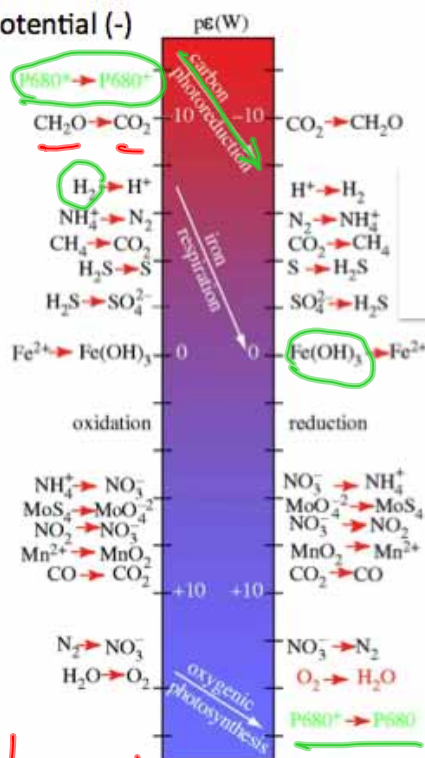
→ Going downhill (low to high potential) generates energy

→ Going uphill (high to low potential) requires energy

Kirschvink and Kopf, Royal Society Trans.

Electron tower

e^- donors
Low potential (-)



A_{red}/A_{ox}

High potential (+)

e^- accep.
 A_{ox}/A_{red}

Thermodynamic underpinning:

From physics, recall:

energy change = electric charge × voltage difference

$$J = C \times V$$

In biology, we can express this as:

Gibbs free energy $\Delta G = -n F \Delta E$

n = number of electrons transferred from reductant (e- donor) to oxidant (e- acceptor)

F = Farady's constant (96,500 C × mol⁻¹)

ΔE = difference in potential between reductant and oxidant

Rule, more ⊖ ΔG, ↑ energy

Week 4,5

CO-EVOLUTION and PHYSICAL AND CHEMICAL FORCES THAT SHAPE LIFE

DKN TAKE HOME MESSAGES

- How do we search for life (continued...)
 - Exhaust all possible abiotic alternative
 - "Geobiology" → Earth ⊕ control
- On a planetary scale, what biosignatures does life leave?
 - O_2 , N_2O
 - photosyn-pigments
- What key principle underpins such biosignatures?
 - chemical disequilibria driven by microbial metabolism → incredible variety of solutions

**CO-EVOLUTION and PHYSICAL AND CHEMICAL FORCES
THAT SHAPE LIFE**

DKN (illustrating using microbes)

- How do we search for life (continued from Tuesday)
- On a planetary scale, what biosignatures does life leave?
- What key principle is behind this?

MMN (illustrating using larger organisms)

- How has the abiotic world shaped form and function of multicellular organisms?

Adaptations to aquatic vs. terrestrial (vs. aerial) existence:



Supporting the body

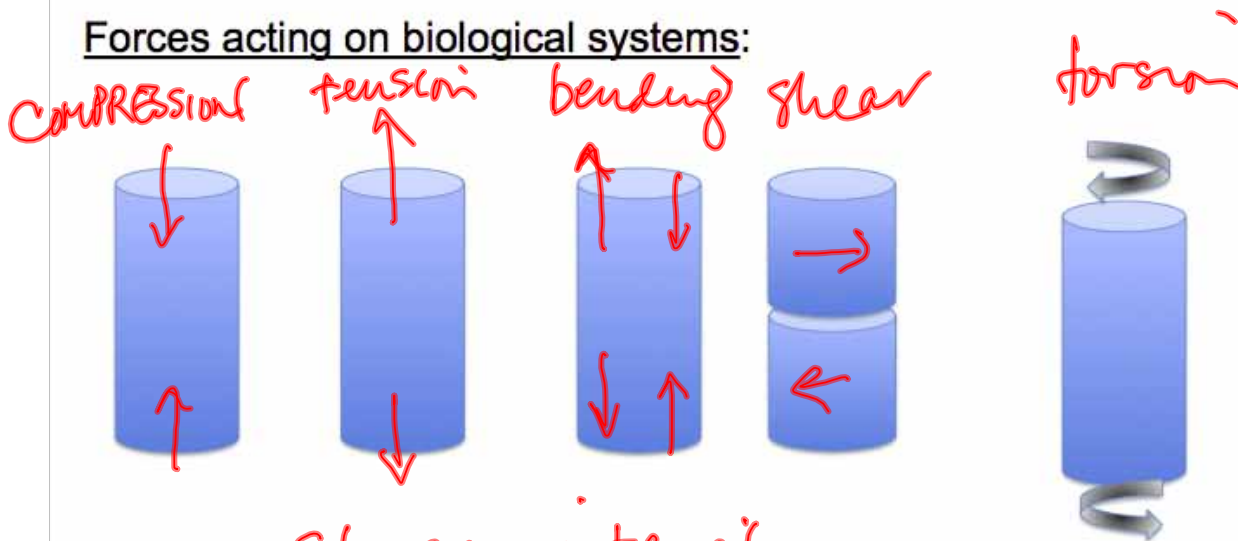
Exchanging O₂ and CO₂



Reproducing

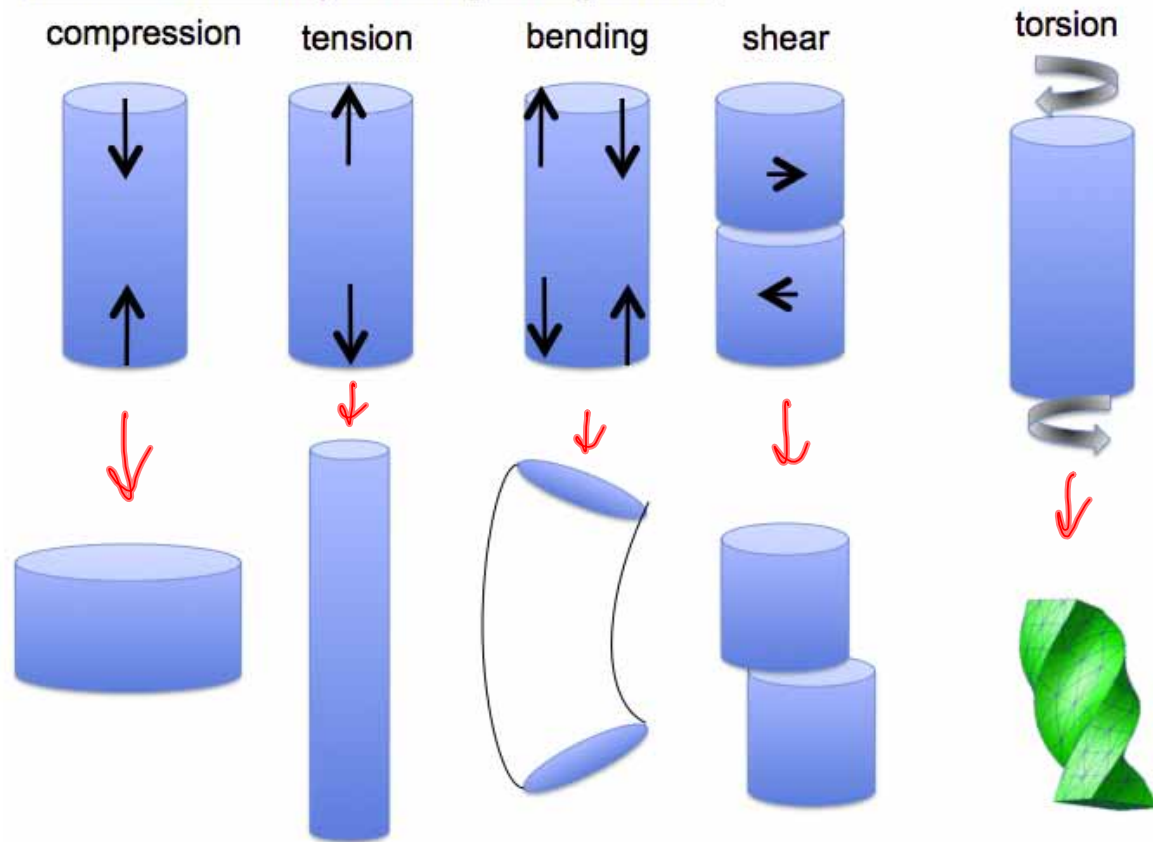
Avoiding desiccation

Forces acting on biological systems:



Stress - intensity
Strain (measurement
of deformation)

Forces that shape biological systems:



Perhaps, the most striking difference between water and air?

Density ($\rho \approx \text{mass / volume}$)
[at sea level, 20°C]

$$\left. \begin{array}{l} \text{H}_2\text{O} / \text{m}^3 = 1000 \text{ kg} \\ * \text{Air} / \text{m}^3 = 1.2 \text{ kg} \end{array} \right\} 830 \times$$

$$\text{Weight} = m \cdot g$$

$$P - P_f$$

$\rho = \text{density}$

ρ_f = density of medium

$$\rho_{eff} = \frac{\rho - \rho_f}{\rho_s - \rho_f}$$

$(\rho - \rho_f) = \text{effective density}$

most biological forms density $\sim$ 1100 kg/m³

~~$\sim 1000 \text{ kg/m}^3$~~ 100 kg/m³ 1000 kg/m³

↓

floats



Compare a pine tree and kelp:

- 20 - 30 m length
- Canopy for photosynthesis w/ distal ~~the~~ ends
- both 1,000 kg / m³



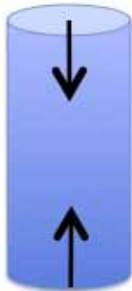
Pine - trunk diameter
0.5 m

kelp stipe 0.05 m

1/20

Forces acting on biological systems:

compression



tension



bending



shear



torsion



Principal forces in air:

*compression
(bending)*

Principal forces in water:

tension (shear)

[mixture]

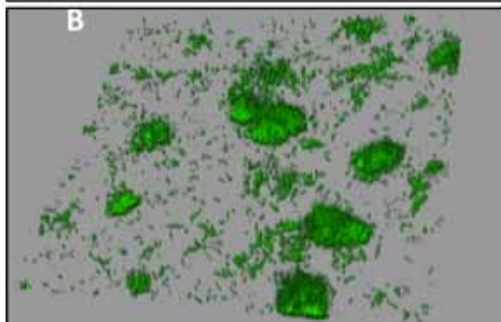
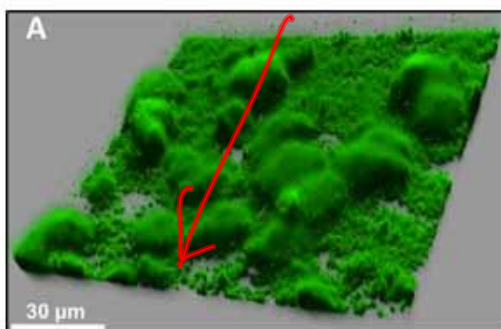
Structural members strained principally in:

1° tension



1° Compression

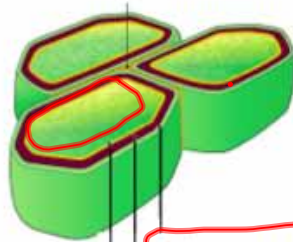




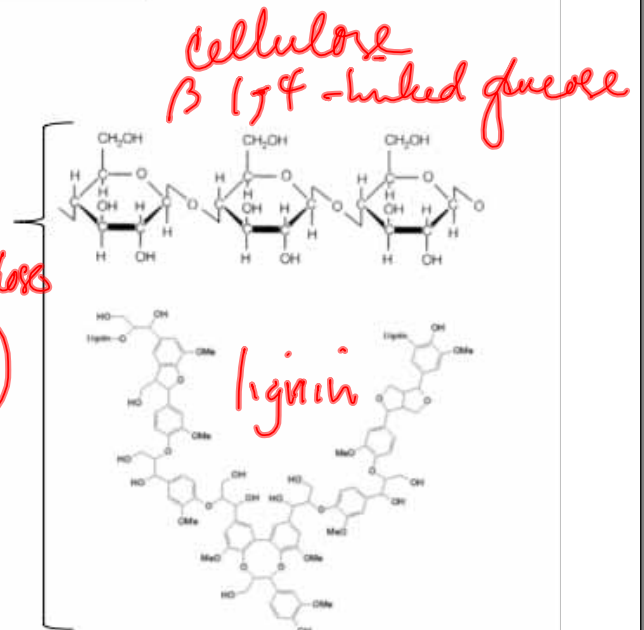
Biofilm
shear



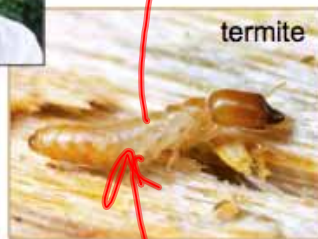
dinoflagellates
shear



hydrocellulose
(biofuels)



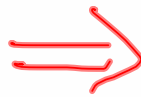
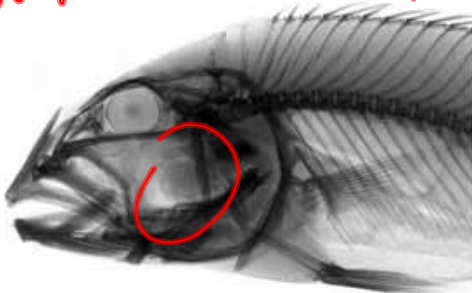
Jared
Leadbetter



termite

30 independent bones
50 muscles

40%



~1%

Animals – structure/function of the skeleton in:

1. feeding

in water – INCOMPRESSIBLE

in air – COMPRESSIBLE (tongue muscles)



2. locomotion

in water – working against medium

on land – working against substrate (gravity)

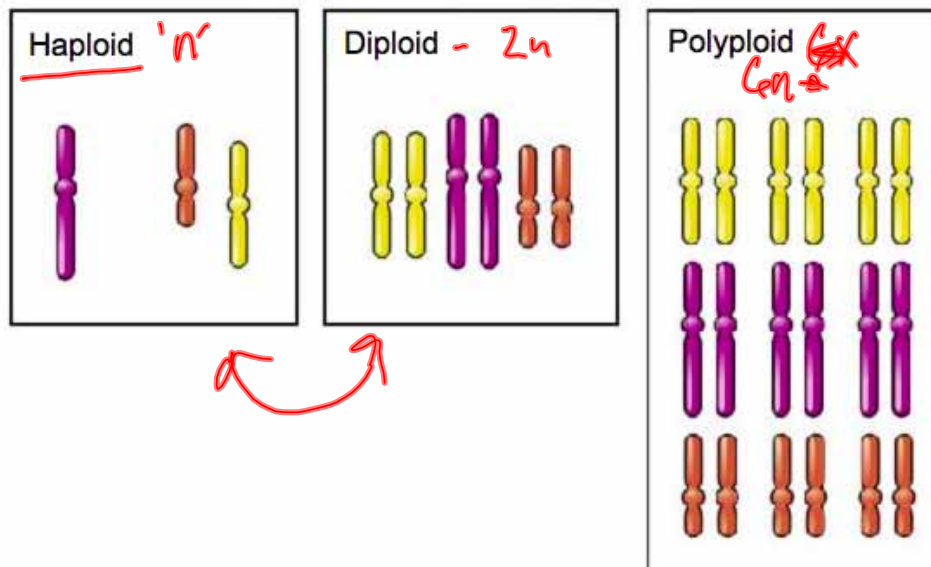
	<u>km/h</u>	<u>body-lengths/sec</u>
Cheetah	<u>112</u>	25
House cat	48	29
Human (Michael Johnson)	45	<u>7</u>
Tiger beetle	1.9	125
[Sailfish	112	25]

Adaptations to aquatic vs. terrestrial (vs. aerial) existence:

- *Supporting the body
- Exchanging O₂ and CO₂
- *Reproducing
- Avoiding desiccation

Reproductive 'Strategy'

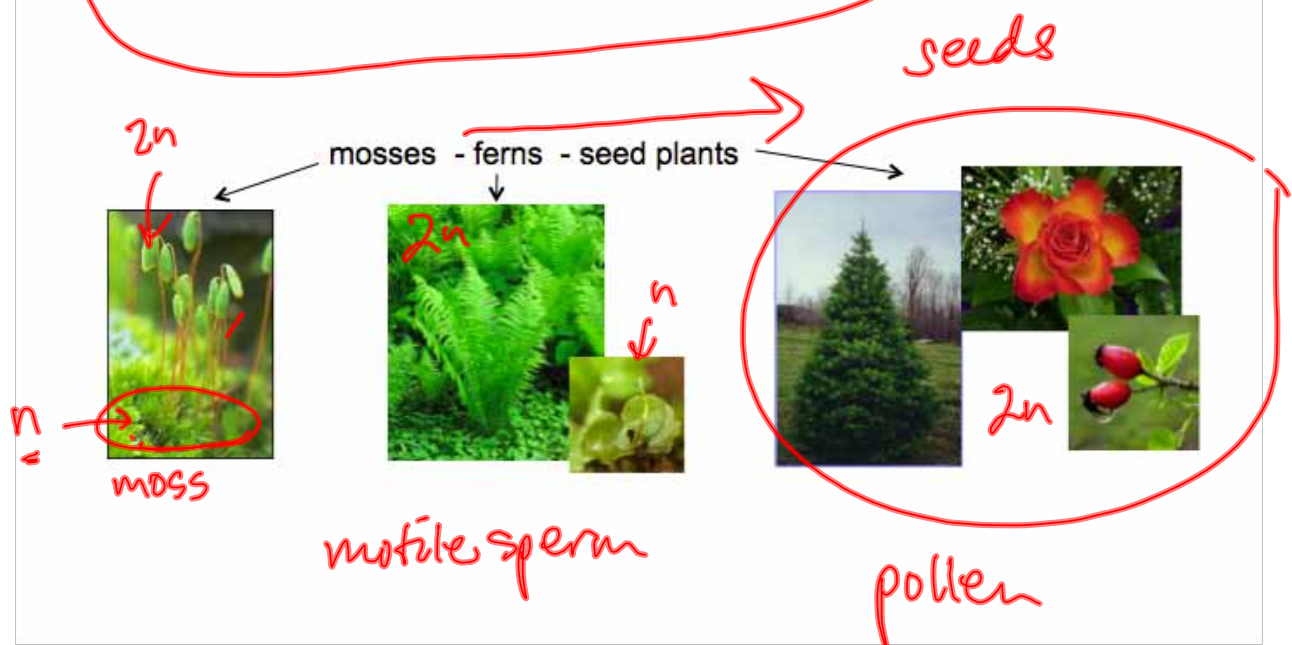
PLOIDY = # OF CHROMOSOME SETS



Evolution in Reproductive 'Strategy' - Plants

In plant evolution:

DOMINANCE 'n' → 2n



Animals – Reproductive Strategy

broadcast

aquatic

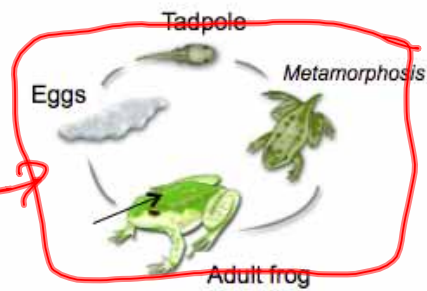


Animals – Reproductive Strategy



'broadcast' spawning

1. *terrestrial – motile sperm*
internal fertilization:
dev. out of H₂O



2. *Shelled EGGS/*
EXTERNAL DEV.



3.



Placental mammals
internal develop

Take-home message:

The presence or absence of an aqueous environment has been a key driver of evolutionary selection pressure on the form and function of multicellular Eukarya.